



**GANDHI SCHOOL OF ENGINEERING
BHABANDHA, BERHAMPUR
SESSION PLAN
3RD SEMESTER, BRANCH-MECHANICAL(GROUP 1)
THERMAL ENGINEERING-1(TH-4)**

Name of the Faculty –ER. SIBASISH SAHU						
Details to be taken				Actually taken		
SL NO & CHAPTER	No. of Periods assigned by SCTE & VT	Details of the topics	PLANNED DATE	Details of the topics	ACTUAL DATE	Remarks
1. Thermodynamic concept & Terminology	12	1.1 Thermodynamic Systems (closed, open, isolated) 1.2 Thermodynamic properties of a system (pressure, volume, temperature, entropy, enthalpy, Internal energy and units of measurement). 1.3 Intensive and extensive properties 1.4 Define thermodynamic processes, path, cycle , state, path function, point function. 1.5 Thermodynamic Equilibrium. 1.6 Quasi-static Process. 1.7 Conceptual explanation of energy and its sources 1.8 Work , heat and comparison between the two. 1.9 Mechanical Equivalent of Heat. 1.10 Work transfer, Displacement work	1/08/2023 TO 22/08/2023	1.1 Thermodynamic Systems (closed, open, isolated) 1.2 Thermodynamic properties of a system (pressure, volume, temperature, entropy, enthalpy, Internal energy and units of measurement). 1.3 Intensive and extensive properties 1.4 Define thermodynamic processes, path, cycle , state, path function, point function. 1.5 Thermodynamic Equilibrium. 1.6 Quasi-static Process. 1.7 Conceptual explanation of energy and its sources 1.8 Work , heat and comparison between the two. 1.9 Mechanical Equivalent of Heat. 1.10 Work transfer, Displacement work	1/08/2023 2/08/2023 4/08/2023 5/08/2023 8/08/2023 9/08/2023 11/08/2023 12/08/2023 16/08/2023 18/08/2023 19/08/2023 22/08/2023	

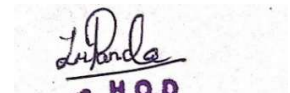
2. Thermodynamics	12	<p>2.1 State & explain Zeroth law of thermodynamics.</p> <p>2.2 State & explain First law of thermodynamics.</p> <p>2.3 Limitations of First law of thermodynamics</p> <p>2.4 Application of First law of Thermodynamics (steady flow energy equation and its application to turbine and compressor)</p> <p>2.4 Second law of thermodynamics (Clausius & Kelvin Planck statements).</p> <p>2.5 Application of second law in heat engine, heat pump, refrigerator & determination of efficiencies & C.O.P (solve simple numerical</p>	<p>23/08/2023 TO 16/09/2023</p>	<p>2.1 State & explain Zeroth law of thermodynamics.</p> <p>2.2 State & explain First law of thermodynamics.</p> <p>2.3 Limitations of First law of thermodynamics</p> <p>2.4 Application of First law of Thermodynamics (steady flow energy equation and its application to turbine and compressor)</p> <p>2.4 Second law of thermodynamics (Clausius & Kelvin Planck statements).</p> <p>2.5 Application of second law in heat engine, heat pump, refrigerator & determination of efficiencies & C.O.P (solve simple numerical</p>	<p>23/08/2023 25/08/2023</p> <p>26/08/2023</p> <p>29/08/2023 1/09/2023 2/09/2023 8/09/2023</p> <p>9/09/2023 12/09/2023</p> <p>13/09/2023 15/09/2023 16/09/2023</p>	
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3. Properties Processes of perfect gas	10	<p>3.1 Laws of perfect gas: Boyle's law, Charle's law, Avogadro's law, Dalton's law of partial pressure, Guy lussac law, General gas equation, characteristic gas constant, Universal gas constant.</p> <p>3.2 Explain specific heat of gas (C_p and C_v)</p> <p>3.3 Relation between C_p & C_v.</p> <p>3.4 Enthalpy of a gas.</p> <p>3.5 Work done during a non- flow process.</p> <p>3.6 Application of first law of thermodynamics to various non flow process (Isothermal, Isobaric, Isentropic and polytrophic process)</p> <p>3.6 Solve simple problems on above.</p> <p>3.7 Free expansion & throttling process.</p>	<p>20/09/2023 TO 6/10/2023</p>	<p>3.1 Laws of perfect gas: Boyle's law, Charle's law, Avogadro's law, Dalton's law of partial pressure, Guy lussac law, General gas equation, characteristic gas constant, Universal gas constant.</p> <p>3.2 Explain specific heat of gas (C_p and C_v)</p> <p>3.3 Relation between C_p & C_v.</p> <p>3.4 Enthalpy of a gas.</p> <p>3.5 Work done during a non- flow process.</p> <p>3.6 Application of first law of thermodynamics to various non flow process (Isothermal, Isobaric, Isentropic and polytrophic process)</p> <p>3.6 Solve simple problems on above.</p> <p>3.7 Free expansion & throttling process.</p>	<p>20/09/2023 22/09/2023</p> <p>23/09/2023</p> <p>26/09/2023 27/09/2023 29/09/2023</p> <p>30/09/2023</p> <p>3/10/2023 4/10/2023 6/10/2023</p>	
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4. Internal combustion engine	8	4.1 Explain & classify I.C engine. 4.2 Terminology of I.C Engine such as bore, dead centers, stroke volume, piston speed &RPM. 4.3 Explain the working principle of 2-stroke & 4- stroke engine C.I & S.I engine. 4.4 Differentiate between 2-stroke & 4- stroke engine C.I & S.I engine.	7/10/2023 TO 1/11/2023	4.1 Explain & classify I.C engine. 4.2 Terminology of I.C Engine such as bore, dead centers, stroke volume, piston speed &RPM. 4.3 Explain the working principle of 2-stroke & 4- stroke engine C.I & S.I engine. 4.4 Differentiate between 2-stroke & 4- stroke engine C.I & S.I engine.	7/10/2023 10/10/2023 11/10/2023 13/10/2023 17/10/2023 18/10/2023 31/10/2023 1/11/2023	
5. Gas Power Cycle	10	5.1 Carnot cycle 5.2 Otto cycle. 5.3 Diesel cycle. 5.4 Dual cycle. 5.5 Solve simple numerical	2/11/2023 TO 18/11/2023	5.1 Carnot cycle 5.2 Otto cycle. 5.3 Diesel cycle. 5.4 Dual cycle. 5.5 Solve simple numerical	2/11/2023 3/11/2023 4/11/2023 7/11/2023 8/11/2023 10/11/2023 11/11/2023 15/11/2023 17/11/2023 18/11/2023	

6. Fuels and Combustion	8	6.1 Define Fuel. 6.2 Types of fuel. 6.3 Application of different types of fuel. 6.4 Heating values of fuel. 6.5 Quality of I.C engine fuels Octane number, Cetane number	21/11/2023 TO 2/12/2023	6.1 Define Fuel. 6.2 Types of fuel. 6.3 Application of different types of fuel. 6.4 Heating values of fuel. 6.5 Quality of I.C engine fuels Octane number, Cetane number REVISION-	21/11/2023 22/11/2023 24/11/2023 25/11/2023 28/11/2023 29/11/2023 1/12/2023 2/12/2023 6/12/2023 8/12/2023	
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 H.O.D
 Mechanical Engineering
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**GANDHI SCHOOL OF ENGINEERING
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3RD SEMESTER, BRANCH-MECHANICAL(GROUP 2)
THERMAL ENGINEERING-1(TH-4)**

Name of the Faculty –ER. SANJAYA KUMAR SAHU						
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Sanjay Kumar Saha

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Lupinda
 H.O.D
 Mechanical Engineering
 Gandhi School of Engg.
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